School of Chemistry University of the Punjab, Lahore Course Outline

Program	ne BS Chemistry	Course Code	Credit Hours 3			
Course Ti						
Course Introduction						
Course introduction This comprehensive course introduces the fundamentals of science, covering its history and various branches. This course will cover the chemical and biochemical aspects of everyday materials and processes, green chemistry, and the chemistry of common-use products. The course also examines energy resources, modern materials like ceramics, plastics, semiconductors, and nanomaterials, as well as the chemistry of food, biofuels, organic farming, and the environment. This course bridges knowledge gaps among students of different scientific areas, providing a well-rounded understanding of science in daily life. Learning Outcomes On the completion of the course, the students will: Understand the basics of chemistry in daily life. Learn the fundamentals of biochemicals. Explore the interdisciplinary nature of material and processes in everyday life. By taking this course, the student will be more accomplished and informative.						
Course Cont ¹ ent Assignments/Readings						
Week 1	Introduction to Science, History and branches of Science, Chemical and physical properties of the materials (i.e., solid, liquid, gases), chemical bonding, types of chemical reactionIntroduction to thermodynamics and kinetics of the chemical reactions, radiation, and spectrum.					
Week 2	Green chemistry. Chemistry of detergents, shampoo, vanishing cream, etc Continue					
Week 3	Energy resources (renewable and non-renewable). Modern materials/chemicals, ceramics, plastics, semiconductors/nanomaterials, cosmetics, etc. Continue					
Week 4	Continue Continue					
Week 5	Food. Biofuels (biomasses) fertilizers/organic farming					
Week 6	Environmental pollution. Global warming & climate change					
Week 7	Continue					

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	Water and plants.				
	Continue				
Week 8	Quiz				
	Midterm exam				
Week 9	Fermentation, Kitchen chemistry and chemistry of baking.				
Week 10	Biomolecules: structure and function of biomolecules, i.e., proteins, lipids, carbohydrates and enzymes, vitamins.Continue				
	Continue				
Week 11	Continue				
	Antioxidants. Medicines. Toxicons.				
Week 12	Continue				
Week 13	Natural hazards and disasters:Earthquake, volcanic eruption, tsunami,Continue				
	Floods, avalanches, drought and salinity,				
Week 14	Continue				
	Wildfire, disaster and risk management.				
Week 15	Continue				
	Presentation				
Week 16	Final term exam				
	Textbooks and Reading Material				
-	Raymond. Physical chemistry for the chemical and biological sciences. University				
	Books, 2000.				
	Nelson, David L., Albert L. Lehninger, and Michael M. Cox. Lehninger principles of biochemistry. Macmillan, 2008.				
3. Enviror	onmental Science: Systems and Solutions By Michael L. McKinney, Robert Schoch				
	ogan Yonavjak, 5th ed. 2013.				
	Edward A., and Duane E. DeVecchio. Natural hazards: earth's processes as s, disasters, and catastrophes. Routledge, 2019.				
5. Calliste	ter Jr, William D., and David G. Rethwisch. Materials science and engineering: an				
	uction. John Wiley & Sons, 2020. nt Muslim Scientists by Sayed Fakhar-e-Alam Naqvi. Peace Publications. 2017.				
	day Science by Dr. Muhammad Akram Kashmiri. A. H. Publishers. 22 AI-Fazal				
	et, Urdu Bazar, Lahore				
-	ay Science by Prof. Dr, Riaz-ul-Haq. Dogar Publishers, 36 Urdu Bazar, Lahore				
	Guide to Science, Isaac Asimov. 1993. https://clalibrary.blogspot.com/2020/06/asimovs-new-guide-to-science-1993-by.html				
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Note:

10. It is preferable to use latest available editions of books. Mention the publisher & year of publication.

11. The References/ bibliography may be in accordance with the typing manual of the concerned faculty/subject. Preferably follow APA 7th Edition publication manual.

Teaching Learning Strategies

- 1. Class lectures
- 2. Quiz
- 3. Assignments
- 4. presentation

Assignments: Types and Number with Calendar

- 1. Assignment # 1: Before mid
- 2. Assignment # 2: Before final

Assessment				
Sr. No.	Elements	Weightage	Details	
1.	Midterm Assessment	35%	Written Assessment at the mid-point of the semester.	
2.	Formative Assessment	25%	Continuous assessment includes: Classroom participation, assignments, presentations, viva voce, attitude and behavior, hands-on-activities, short tests, projects, practical, reflections, readings, quizzes etc.	
3.	Final Assessment	40%	Written Examination at the end of the semester. It is mostly in the form of a test, but owing to the nature of the course the teacher may assess their students based on term paper, research proposal development, field work and report writing etc.	